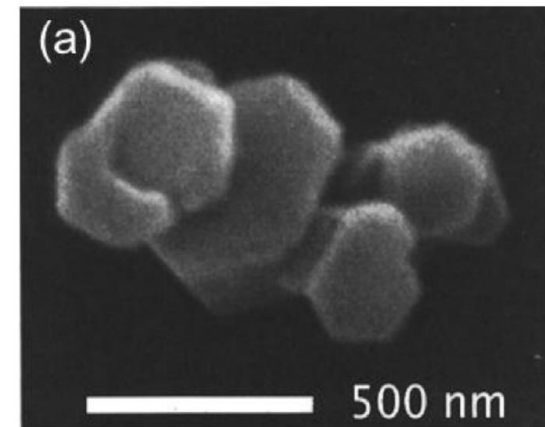
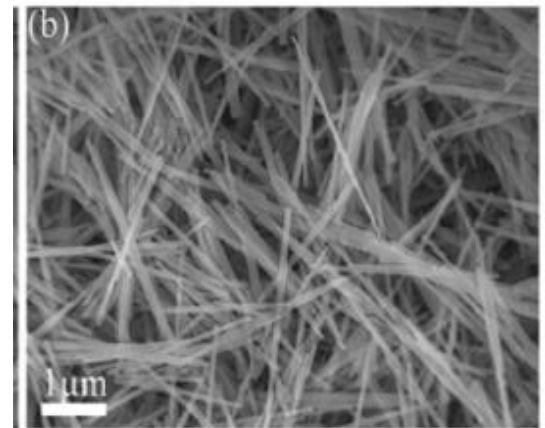
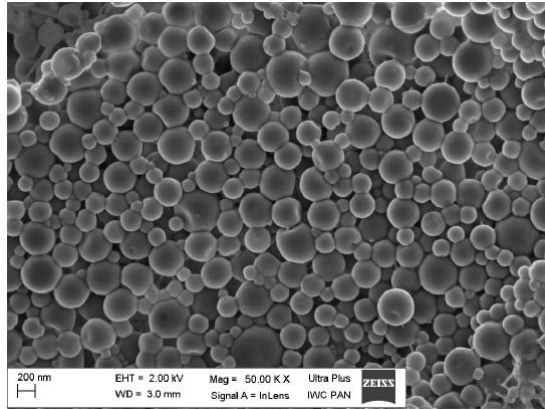


CaO

Nanomateriali nell'industria dei cementi e per la sostenibilità ambientale

Paola Comodi

Dipartimento di Fisica e Geologia
Università di Perugia



Perché i cementi ?

concrete is

the most widely used **substance** on the planet **after water!!!!**

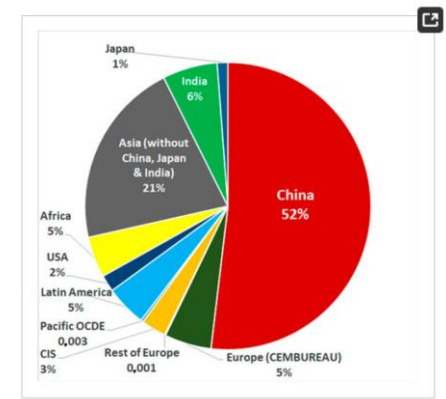
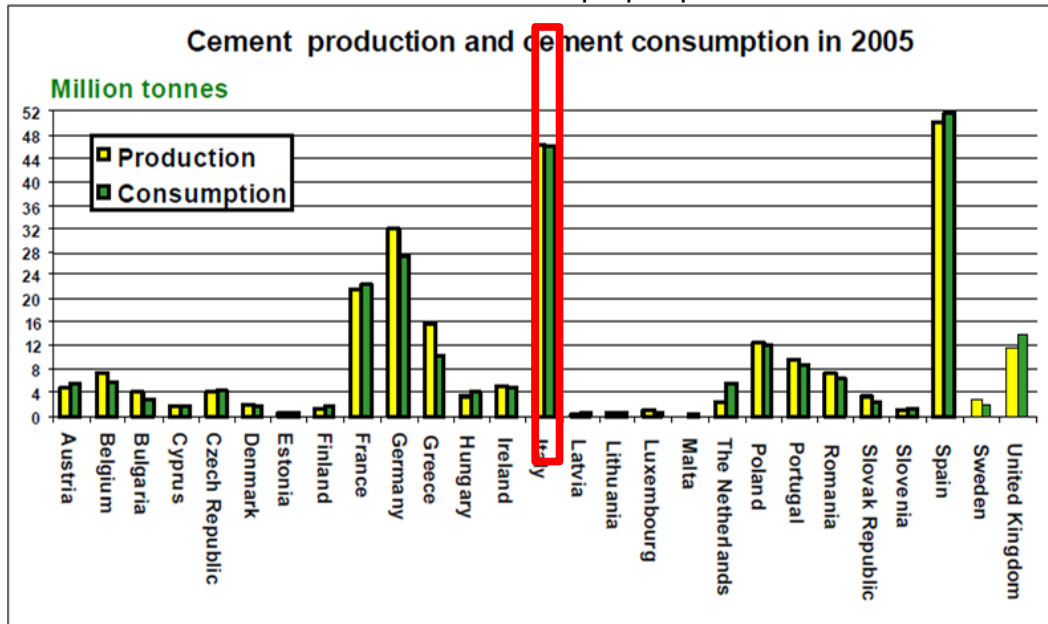


Figure 2. World cement production in 2016 by region and main countries (4.65 billion tons) [10].

ITALIA → **Seconda** nazione in Europa per produzione e consumo di cemento



ITALIA → Tra le **18 nazioni al mondo** per produzione di cemento e portata di clinker (USGS Mineral Program Cement Report - February 2014)

World Production and Capacity:

	Cement production		Clinker capacity ^e	
	2012	2013 ^e	2012	2013
United States (includes Puerto Rico)	74,900	77,800	106,000	105,000
Brazil	68,800	70,000	57,000	60,000
China	2,210,000	2,300,000	1,800,000	1,900,000
Egypt	46,100	46,000	46,000	46,000
Germany	32,400	34,000	31,000	31,000
India	270,000	280,000	270,000	280,000
Indonesia	32,000	35,000	47,500	50,000
Iran	70,000	75,000	75,000	80,000
Italy	33,000	29,000	46,000	46,000
Japan	51,300	53,000	55,000	55,000
Korea, Republic of	48,000	49,000	50,000	50,000
Mexico	35,400	36,000	42,000	42,000
Pakistan	32,000	32,000	42,500	42,000
Russia	61,500	65,000	80,000	80,000
Saudi Arabia	50,000	50,000	55,000	55,000
Thailand	37,000	35,000	50,000	50,000
Turkey	63,900	70,000	66,900	67,000
Vietnam	60,000	65,000	68,000	70,000
Other countries (rounded)	524,000	597,000	312,000	291,000
World total (rounded)	3,800,000	4,000,000	3,300,000	3,400,000

Perché i cementi ?

Infrastrutture ultrahigh performanti:
aumento di

- Funzionalità
- Caratteristiche meccaniche,
- Resistenza a trazione
- Resistenza a sforzi di taglio
- Resistenza all'alterazione

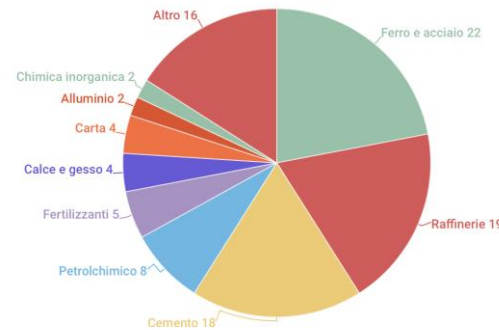
New global standards:

Cementi green che riducano il consumo di energia e la produzione di CO₂

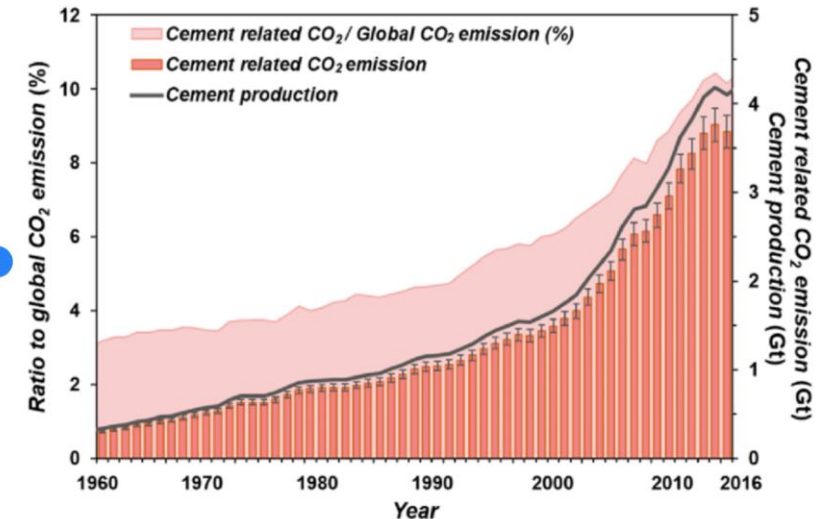
Dall-8-10 % della CO₂ di origine antropica nel mondo viene dalla manifatture del cemento

Rilasciata sia per il combustibile (220 kg di carbone per 1 ton cemento) sia per la de-carbonatazione delle materie prime

Grattacieli più alti del mondo



Percentuale di CO2 prodotta dai vari settori industriali in Europa



CO2 emissions related to the production of cement and its ratio to total CO2 emissions. Reproduced with permission from Thomas D. Kelly and Grecia R. Matos, Historical statistics for mineral and material commodities in the U.S.; published by the United States Geological Survey (USGS), 2015 [85].

Cementi rispondenti al new global standard

Che siano. ...

Più performanti però.....

....Che producano meno CO_2

....Che consumino meno materie prime

Come fare ????

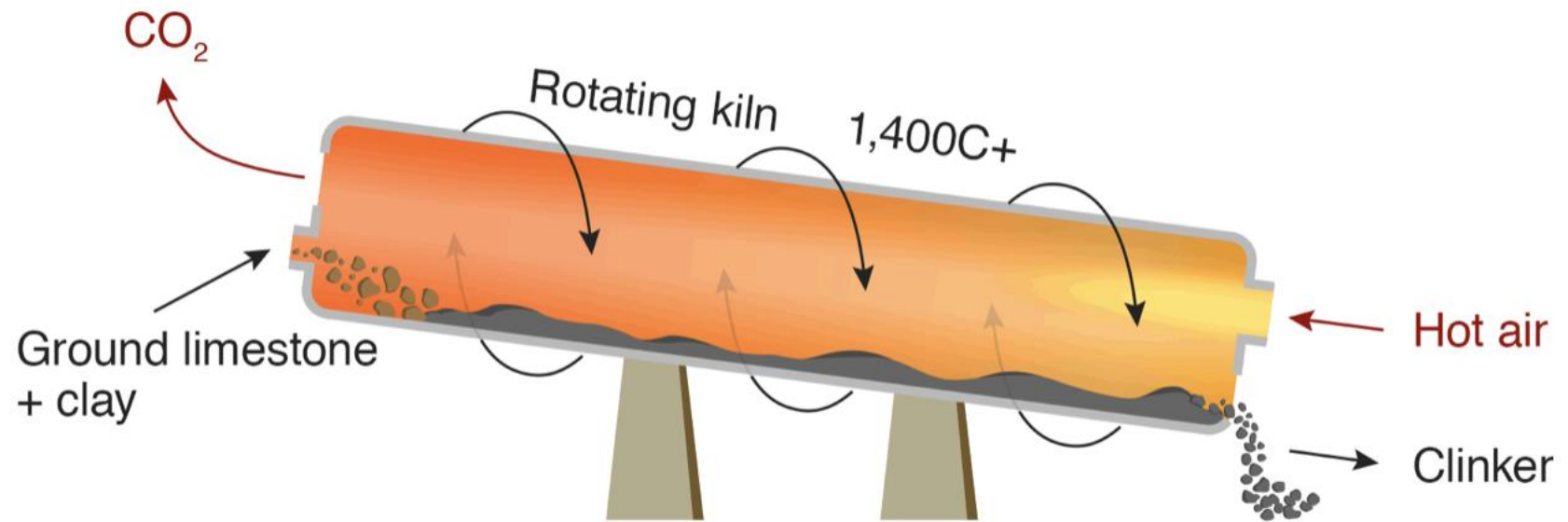
Cosa sono i cementi ?

Cemento Portland

Le materie prime

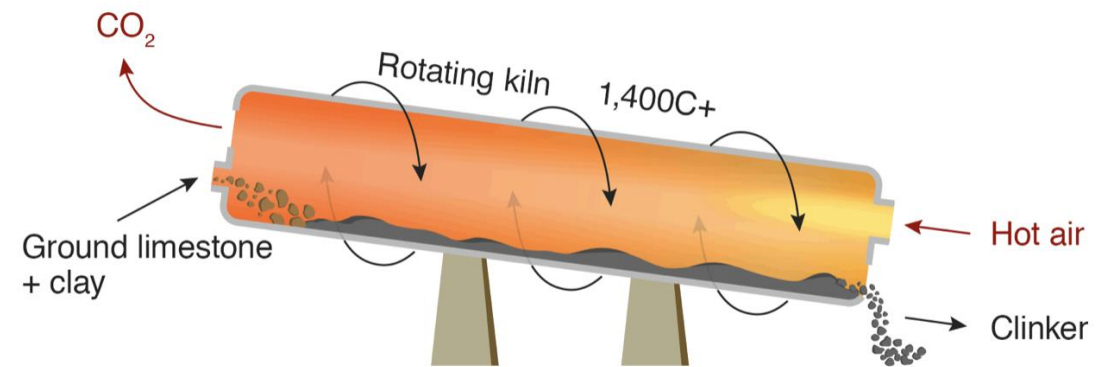
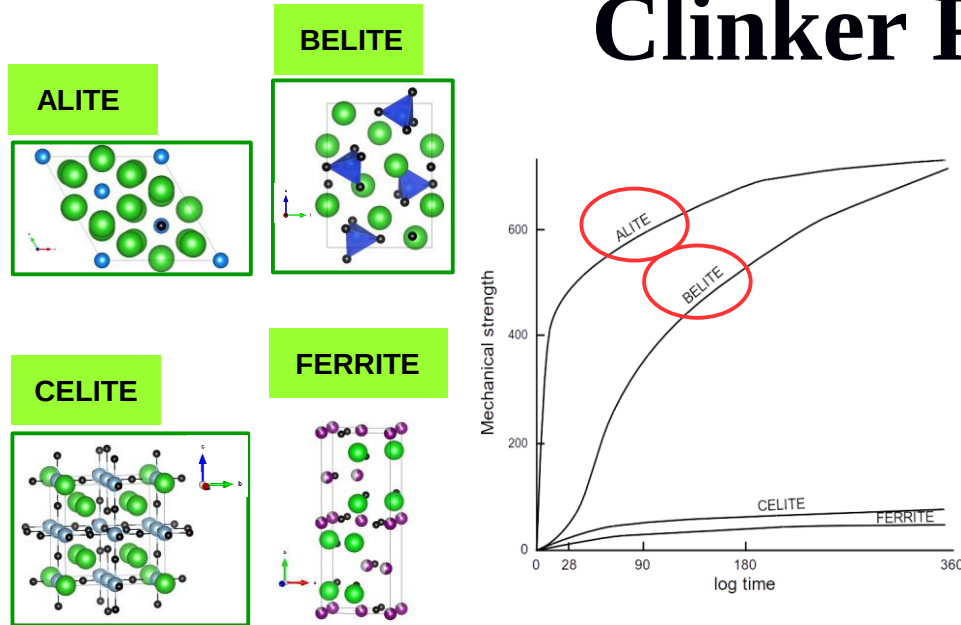


Calcari ed argille



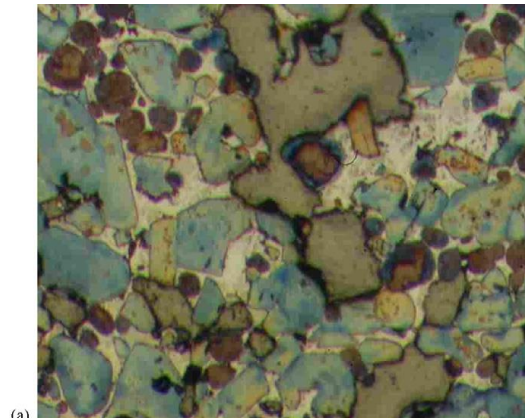
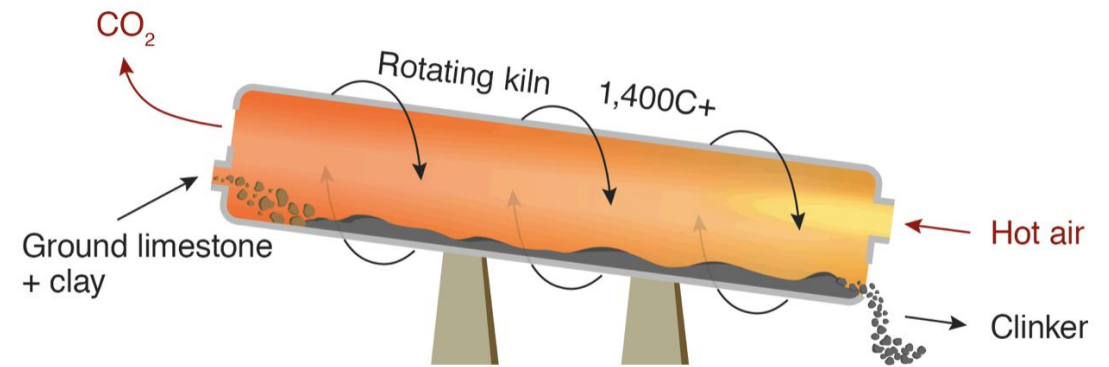
Cosa è il clinker?

Clinker Portland

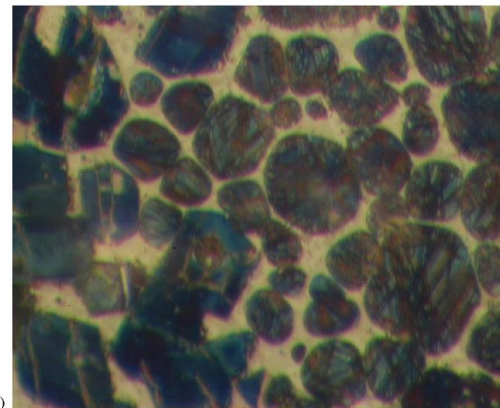


Formula	Ossidi	Abbreviazione	Nome	% _p
Ca_3SiO_5	$3\text{CaO} \cdot \text{SiO}_2$	C3S	Alite	50-70
Ca_2SiO_4	$2\text{CaO} \cdot \text{SiO}_2$	C2S	Belite	13-30
$\text{Ca}_3\text{Al}_2\text{O}_6$	$3\text{CaO} \cdot \text{Al}_2\text{O}_3$	C3A	Celite	5-10
$\text{Ca}_4\text{Al}_2\text{Fe}_2\text{O}_{10}$	$4\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot \text{Fe}_2\text{O}_3$	C4AF	Ferrite	5-15

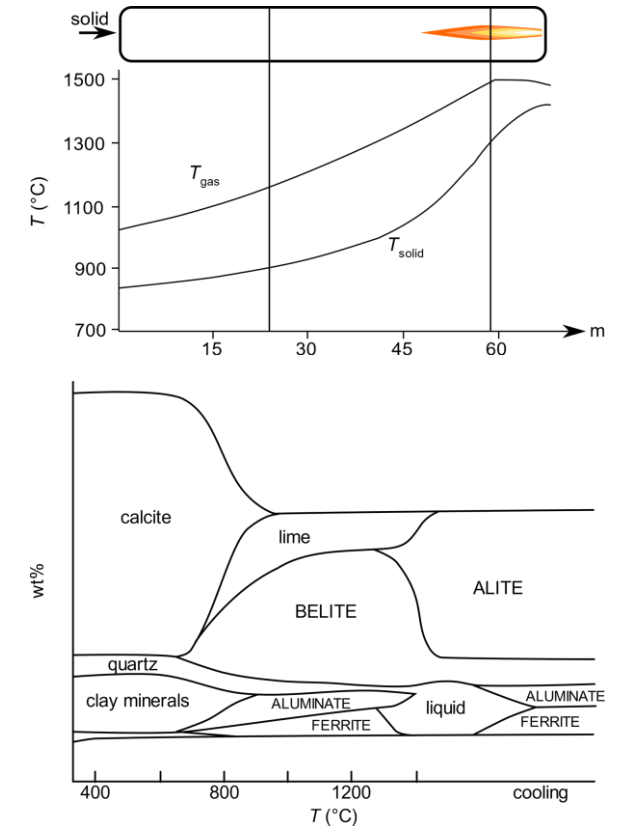
Produzione clinker



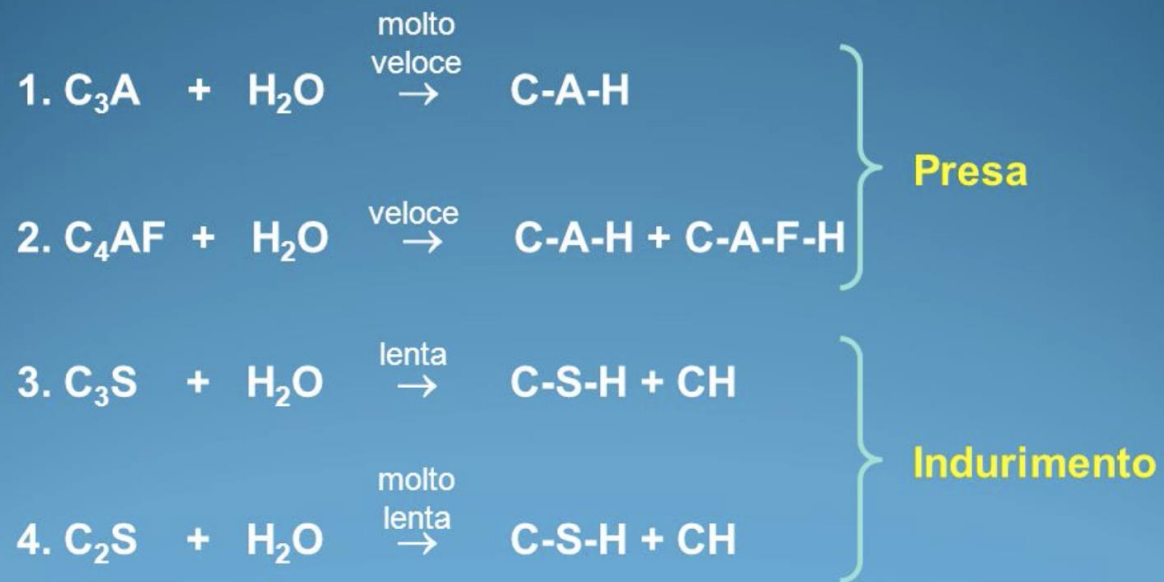
(a)



(b)



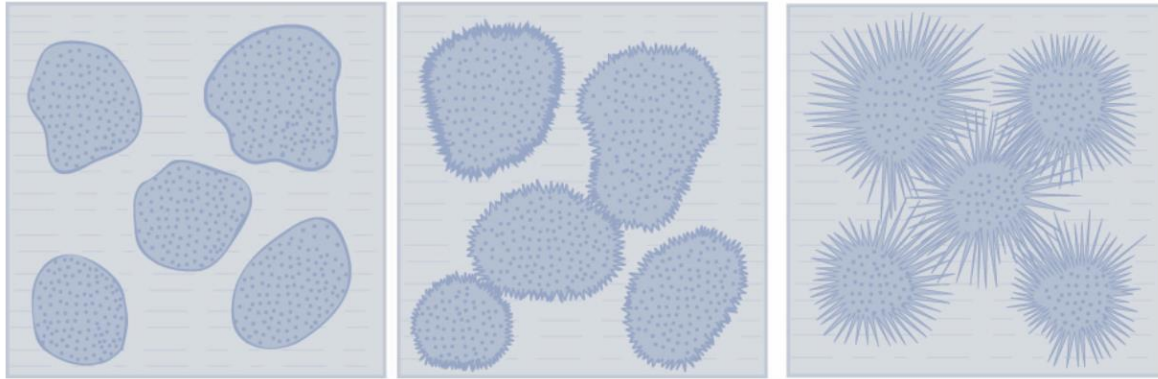
Idratazione del Cemento Portland



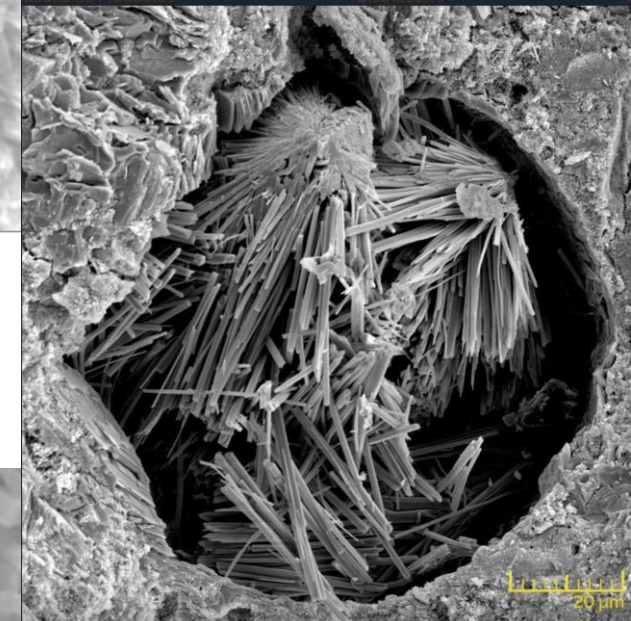
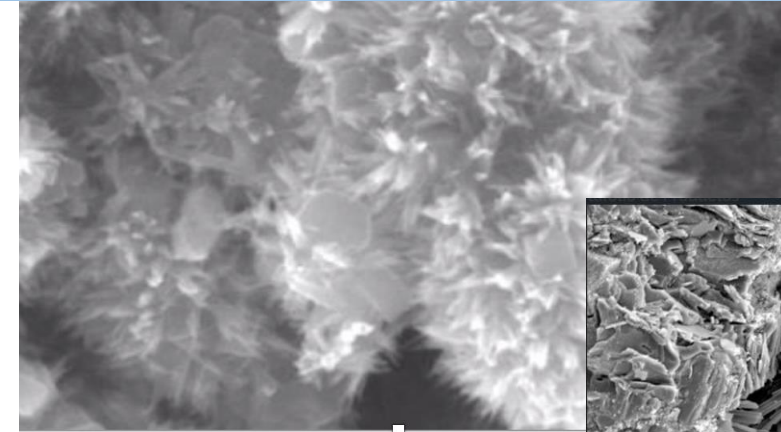
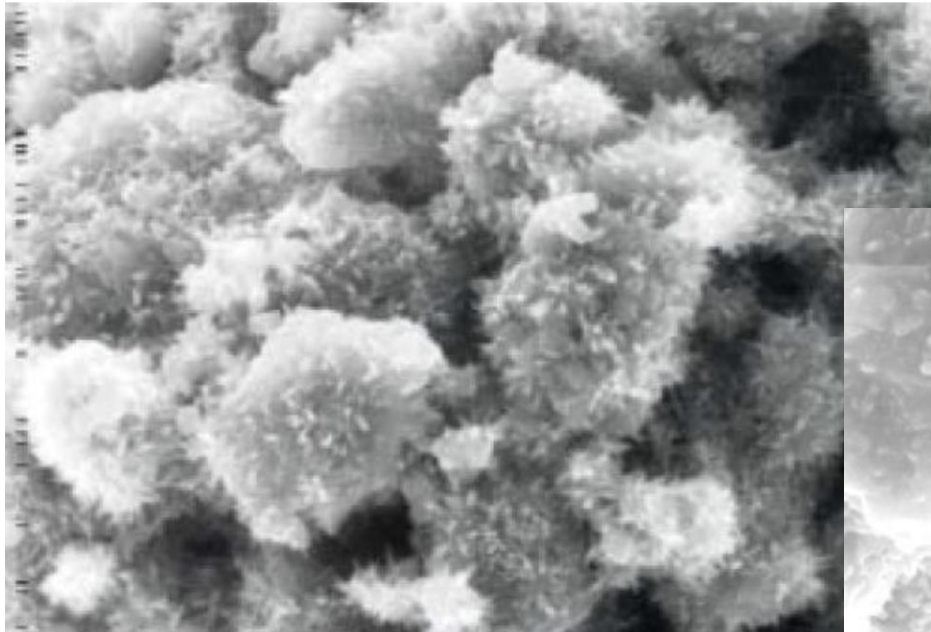
+



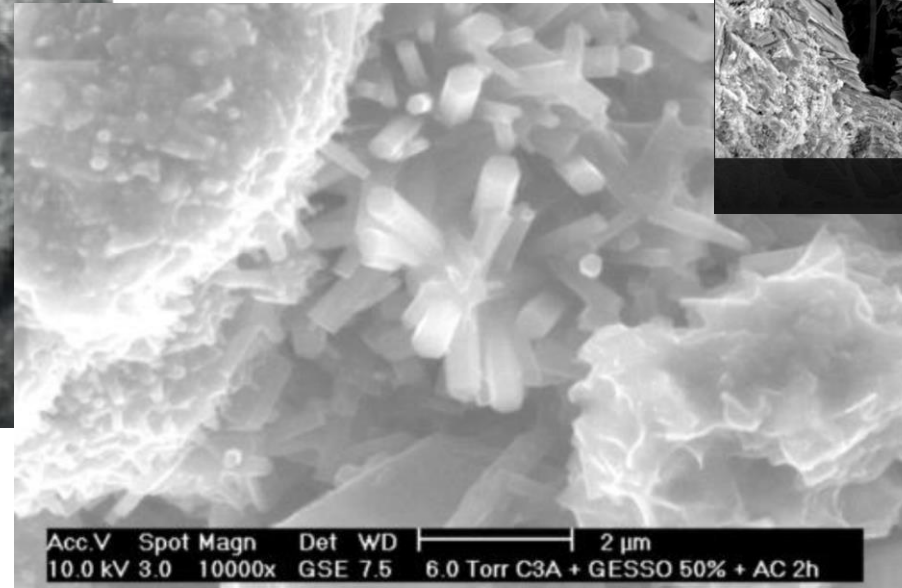
L'idratazione del clinker C-S-H



1a (a), inizio della presa (b) e durante l'indurimento (c)

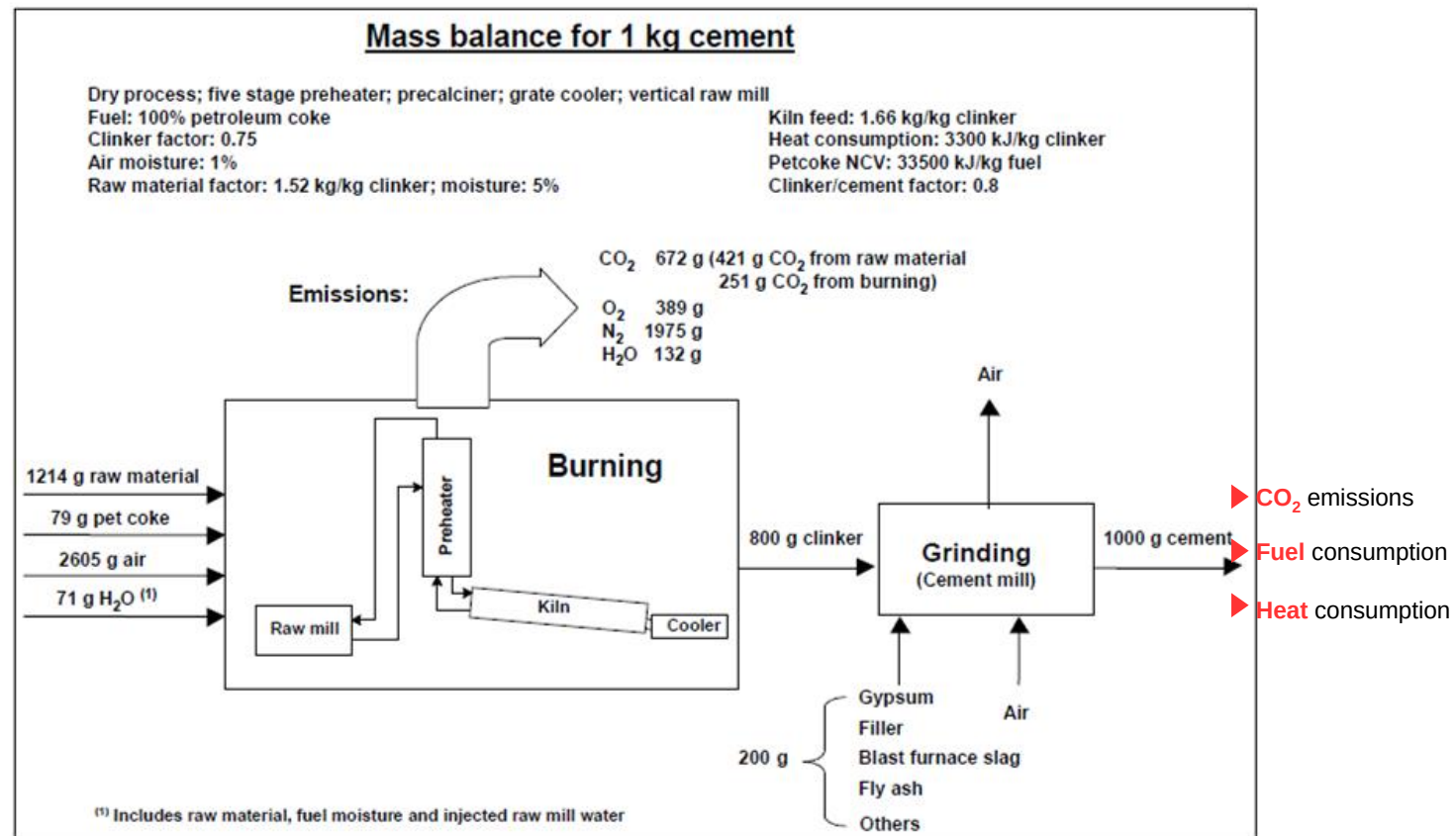


Ettringit in concrete



Acc.V Spot Magn Det WD | 2 μm
10.0 kV 3.0 10000x GSE 7.5 6.0 Torr C3A + GESSO 50% + AC 2h

- Wastes as raw materials
- Non-potable water
- Dry processes



Mass balance for the production of 1 kg cement using the dry process [103, CEMBUREAU, 2006]

“Reference Document on Best Available Techniques in the Cement, Lime and Magnesium Oxide Manufacturing Industries” European Commission - May 2010

Come fare cementi per i.... new global standards

Cementi attivati con :

Scaglie ferrose

Fly ashes

Fumo di silice

Alluminati Ca

Nanomateriali



sustainability

Article

Use of Iron and Steel Slags in Concrete: Future Perspectives

Alan Piemonti ¹, Antonio Conforti ^{1,*}, Luca Comin ¹ and Giovanni Plizzari ¹

Expansive Cement

Expansive cement
of drying shrinkage
restrained expansion

Calcium Aluminate Cements

Karen L. Scrivener, Alain Capmas, in *Lea's Chemistry of Cement and Concrete (Fourth Edition)*, 1998

CONCRETE

Optimizing the Use of Fly Ash in Concrete

By: Michael Thomas, Ph.D., P.Eng., Professor of Civil Engineering, University of New Brunswick



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Construction and Building Materials

Volume 276, 22 March 2021, 122236



Experimental Investigation on Strength and Durability of Graphene Nanoengineered Concrete

Sejal P. Dalal ^{a,*}, Purvang Dalal ^b

^a Civil Engineering Department, SVIT, Gujarat, India

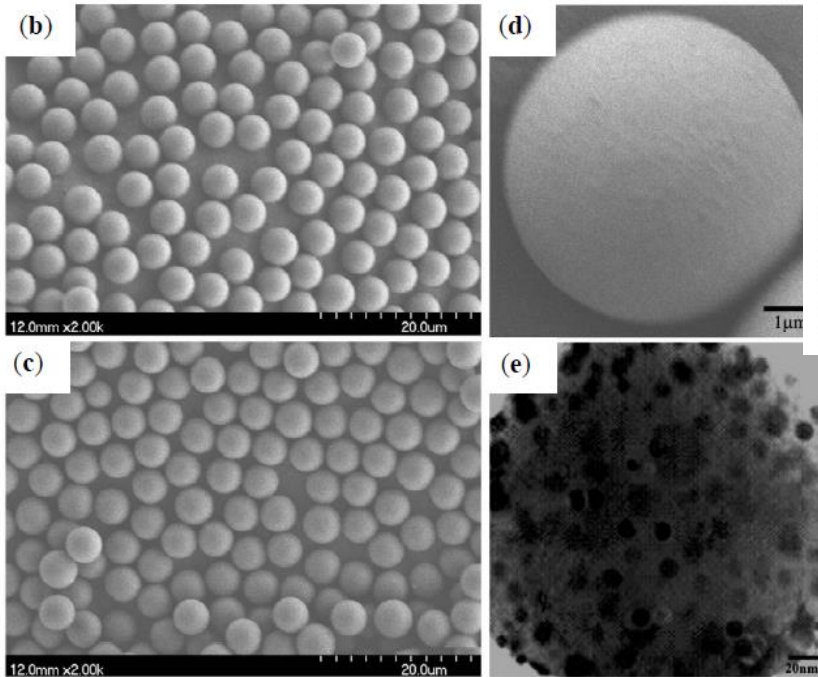
^b Faculty of Technology, Dharmsinh Desai University, Nadiad, Gujarat, India

Experimental Investigation on Strength and Durability of Graphene Nanoengineered Concrete

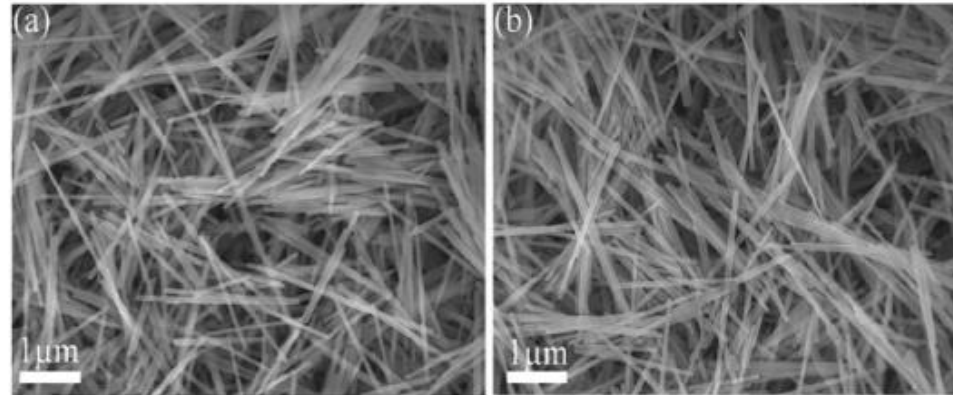
Sejal P. Dalal ^a, Purvang Dalal ^b

Cementi nano-ingegnerizzati

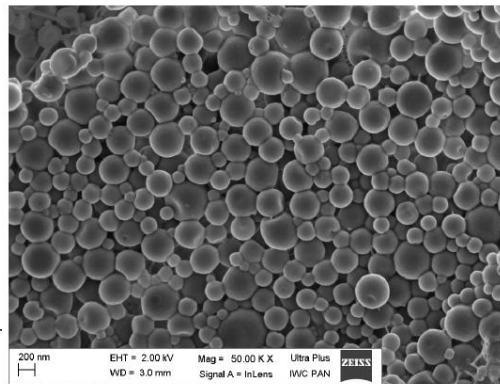
- Materials with a grain size **lower than 200nm**



Nguyen et al., 2014, *Nanomaterials*, 4(3), 628 –



Halayqa & Domańska, 2014, *Int. J. Mol. Sci.*, 15(12), 23909 – 23923



Shakir et al., 2014, *RSC Advances*, 4, 8741 – 8745



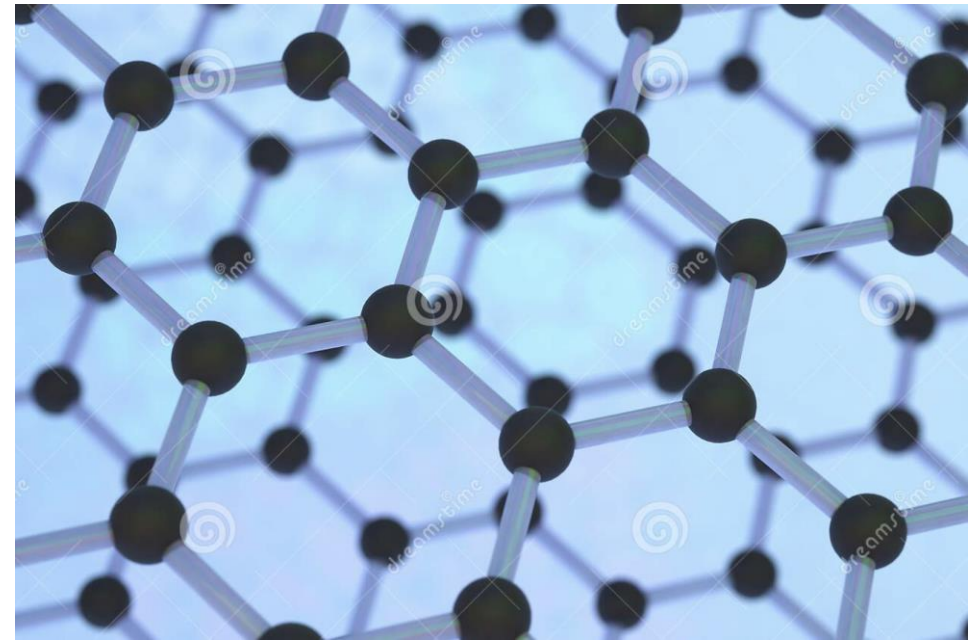
aumentare la durabilità e le proprietà meccaniche
Ridurre la porosità
Facilitare la formazione dei C-S-H

Ultrahigh Performance Nanoengineered Graphene–Concrete Composites for Multifunctional Applications

Dimitar Dimov, Iddo Amit, Olivier Gorrie, Matthew D. Barnes, Nicola J. Townsend, Ana I. S. Neves, Freddie Withers, Saverio Russo, Monica Felicia Craciun✉

First published: 23 April 2018 | <https://doi.org/10.1002/adfm.201705183> | Citations: 79

The copyright line of this paper was changed on 9 August 2018 after initial publication.



Ultrahigh performance multifunctional construction materials.....

Nanoengineered concrete con graphene

+146% resistenza compressione

+ 79.5 resistenza alla flessione/trazione

Aumento performance termiche ed elettriche

+ 400 % permeabilità all'acqua

Cosa sono i nCaO e nCa(OH)₂

Dimensioni particelle 50-250 nanometri

Stabili se disperse in alcoli

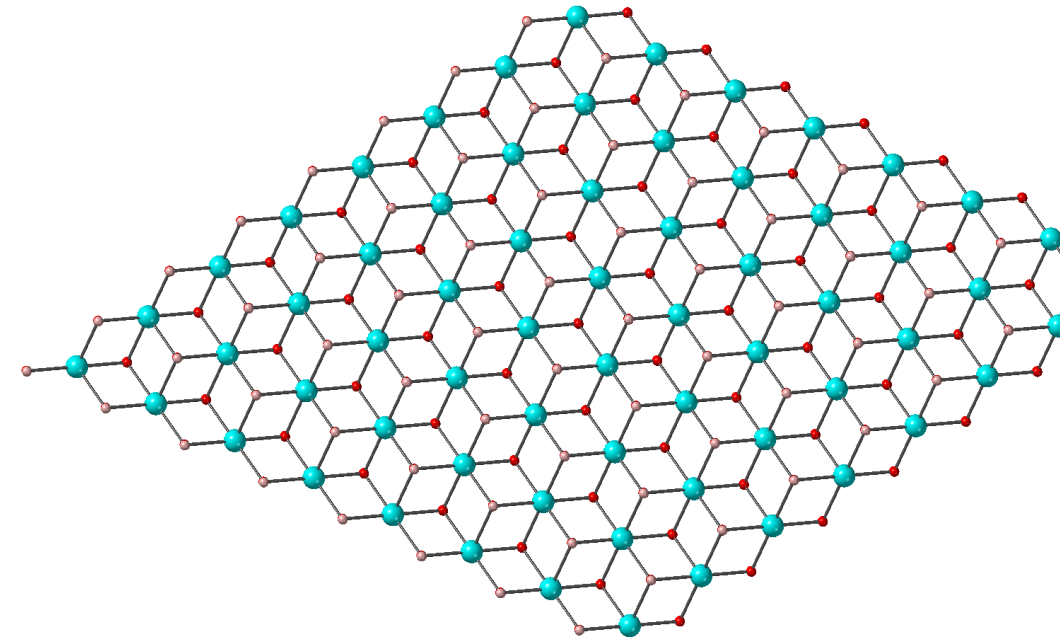
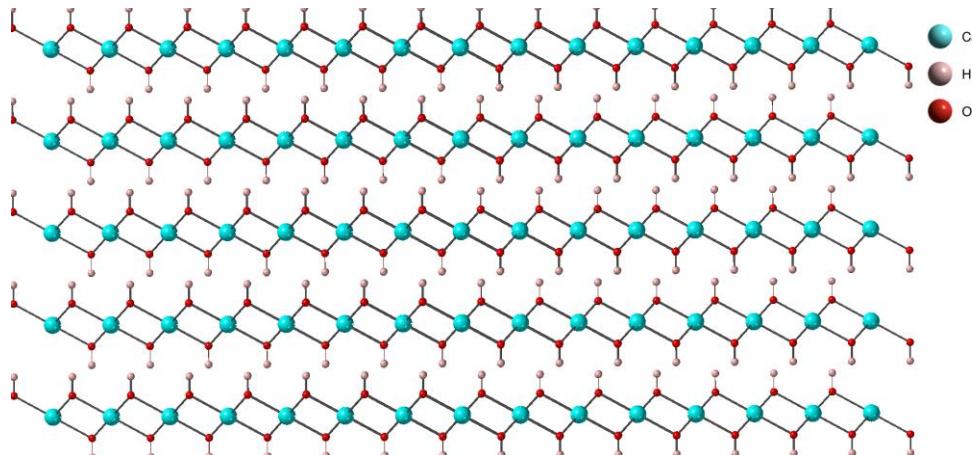
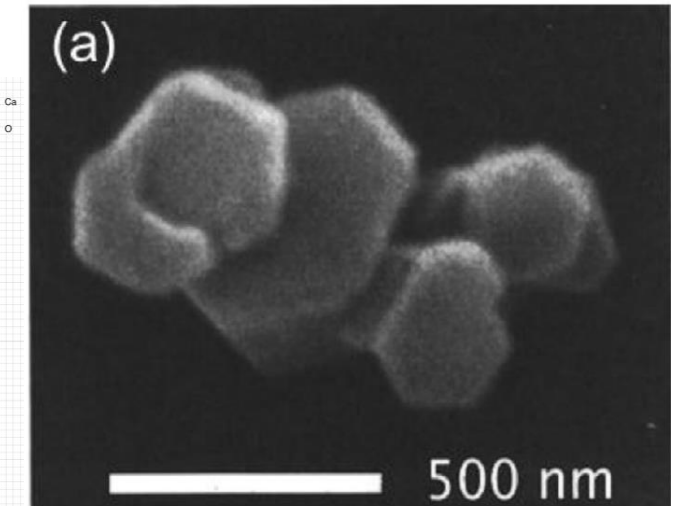
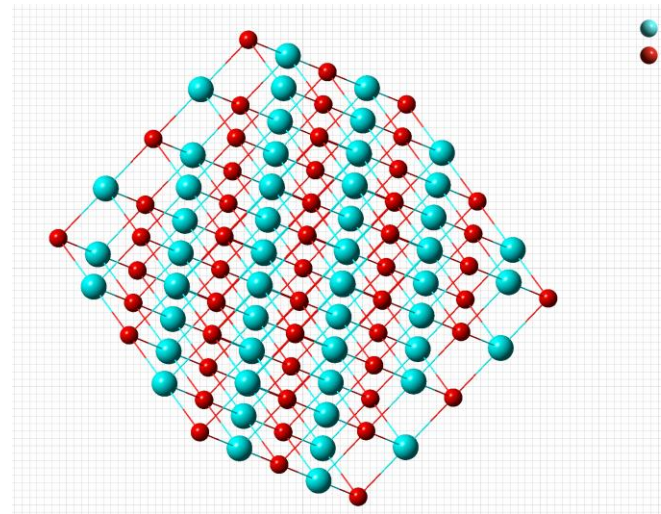
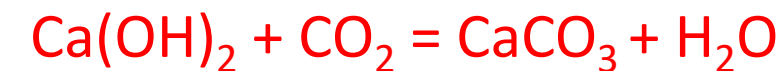
Si formano dalla decarbonatazione



Tendono ad idratarsi a formare nanoportandite



Riconverte facilmente in



A cosa serve il nCaO

Per il restauro e la conservazione delle malte e negli stucchi archeologiche a matrice carbonatica....

Riconverte facilmente in CaCO_3 $\text{Ca(OH)}_2 + \text{CO}_2 = \text{CaCO}_3 + \text{H}_2\text{O}$

Rispetto al trattamento tradizionale: completamento carbonatazione, profondità di penetrazione e minore acqua

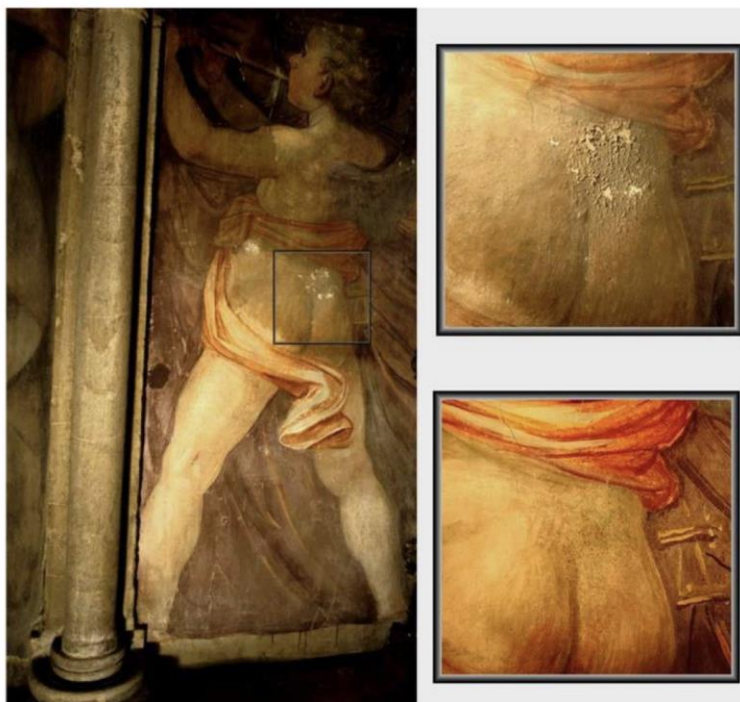
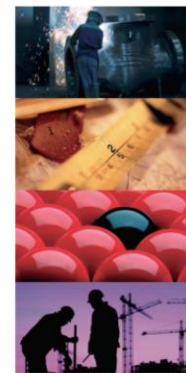


Figura 2.1.1. Affresco "Gli angeli musicanti" di Santi di Tito (16° sec.) sulla controfacciata del Duomo di Firenze. L'area selezionata è stata trattata con nanoparticelle di Ca(OH)_2 (in alto prima del trattamento, in basso dopo il trattamento) [Baglioni, P., "Tecnologie del futuro per i tesori del passato. Le nanoscienze e la conservazione dei beni culturali", Dipartimento di Chimica e CSGI, Università di Firenze, Sesto Fiorentino, 2006].



Figura 2.1.3. Dipinti Maya scoperti nella "Antigua Ciudad Maya de Calakmul", patrimonio mondiale dell'UNESCO (Campeche, Messico). Dispersioni di nanoparticelle di Ca(OH)_2 sono state utilizzate per consolidare lo strato dipinto decoeso e soggetto a polverizzazione. Dopo il restauro la superficie dipinta ha recuperato la sua tonalità di colore originale perché la ri-coesione dei pigmenti nello strato superficiale minimizza la diffusione della luce che aveva conferito opacità alle pitture murali [P. Baglioni, R. Giorgi, *Soft Matter* (2006) 2, 293].



L'impiego di nanotecnologie e nanomateriali per il recupero e la conservazione dei beni culturali

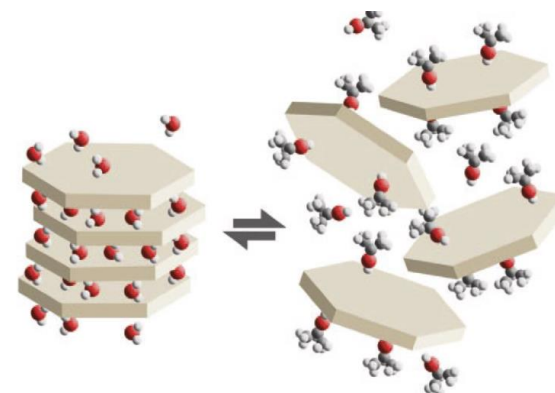
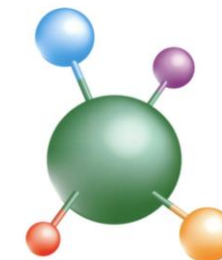


Figura 2.1.2. Sx: aggregazione delle nanoparticelle di idrossido di calcio in acqua; Dx: l'adesione delle facce è inibita grazie all'assorbimento del 2-propanolo, portando alla stabilizzazione della dispersione [P. Baglioni, D. Chelazzi, R. Giorgi, In: "Nanotechnologies in the Conservation of Cultural Heritage A compendium of materials and techniques", Springer (2015)].

A cosa serve il nCaO

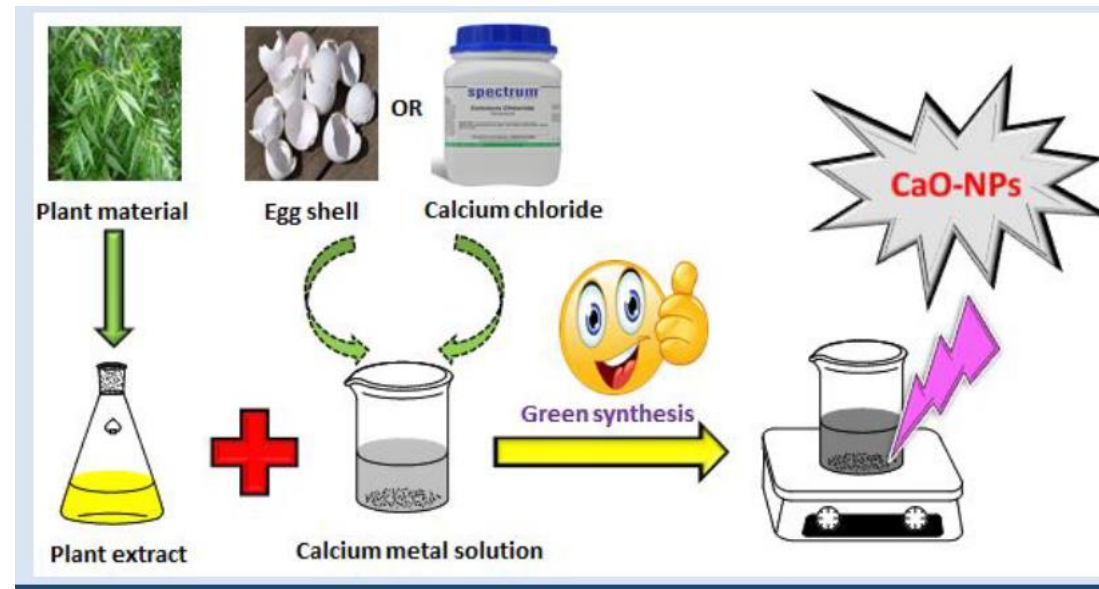
PREPARATION AND CHARACTERIZATION OF CaO NANOPARTICLE FOR BIODIESEL PRODUCTION

Jharna Gupta^{1,b)} and Madhu Agarwal^{1,a)}

¹Department of Chemical Engineering, MNIT, Jaipur, 302017, India

^{a)}Corresponding author: madhunaresh@gmail.com

^{b)}onlinejharna@gmail.com



Tahvildari et al. *Journal of Environmental Health Science & Engineering* (2015) 13:73
DOI 10.1186/s40201-015-0226-7



JOURNAL OF
ENVIRONMENTAL HEALTH
SCIENCE & ENGINEERING

RESEARCH ARTICLE

Open Access

The study of CaO and MgO heterogenic nano-catalyst coupling on transesterification reaction efficacy in the production of biodiesel from recycled cooking oil



Kambiz Tahvildari¹, Yasaman Naghavi Anaraki¹, Reza Fazaeli², Sogol Mirpanji¹ and Elham Delrish^{3*}

nCaO nei cementi per ridurre emissioni CO₂

65% delle emissioni di CO₂ del settore cementi viene dalla calcinazione



$$1 \text{ kg} \rightarrow 0.56 \text{ kg} + 0.44 \text{ kg}$$

In un 1 kg di clinker ci sono 0.65 kg di CaO

35 % delle emissioni CO₂ viene dalla combustione

Cosa ci si può aspettare dal nCaO

Accelerare le reazioni con minore energia data la elevate superficie di reazioni

Aumentare la cuocibilità

Diminuire le temperature di reazione

Aumentare la resistenza e diminuire la compressione

Aumentare l'attività pozzolanica

Diminuire la porosità

Calcare + Argilla + Nano-materiale

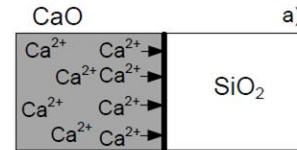


↓ Dimensione

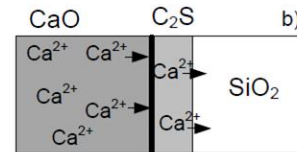
↑ Energia superficiale

↓ T di fusione

↑ Coefficiente di diffusione



Diffusione di Ca^{2+}



nano-CaO

Sostituzione di parte dei carbonati con CaO → **diminuzione delle emissioni di CO_2**

Aumento della diffusione di Ca^{2+} **Diminuzione della T di clinkerizzazione**

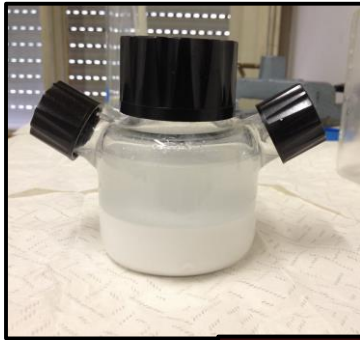
Quali nCaO?

The formation of the hydroxide particles occurs at the interface between the bubble and the liquid and the ultrasound accelerated the formation of $\text{Ca}(\text{OH})_2$.
(Alavi & Morsali, 2013, *J. Exp. Nanosci.*, 5(2), 93 – 105)

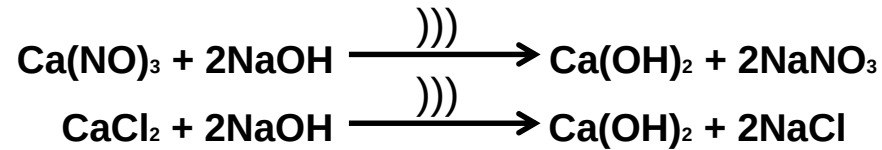
Nano-materials

Synthesis of nano-CaO by acoustic cavitation

Nano-CaO

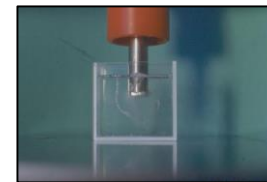
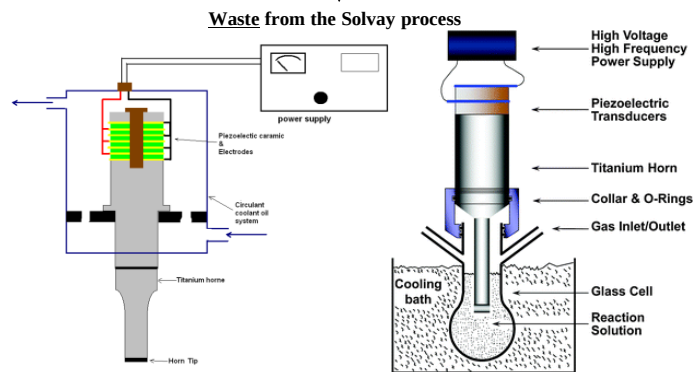


From Calcium Nitrate and Calcium Chloride

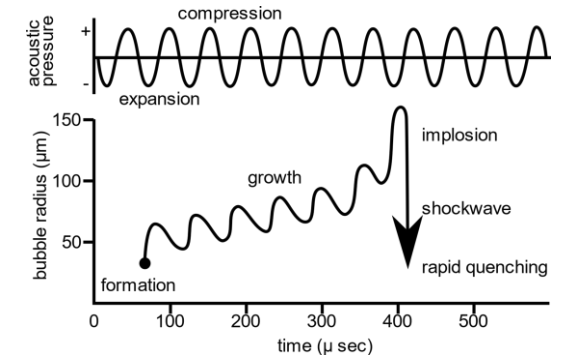
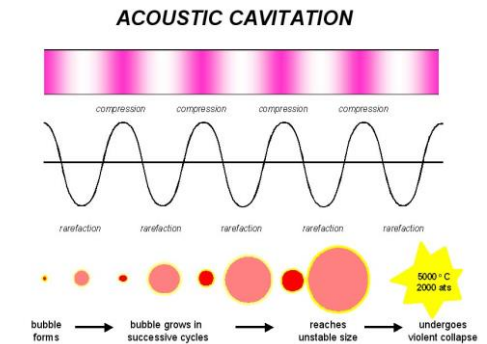


Possible re-agglomeration after calcination

NO calcination $\rightarrow$ **$\text{Ca}(\text{OH})_2$**



Ultrasound Generator:
Horn type 750 W



Suslick, 1990, *Science*, 247,1439 – 1445

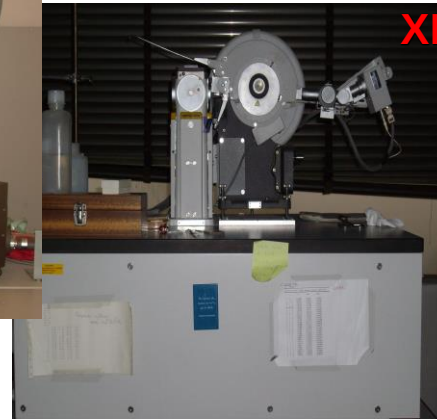
Il clincker nei nostri labs



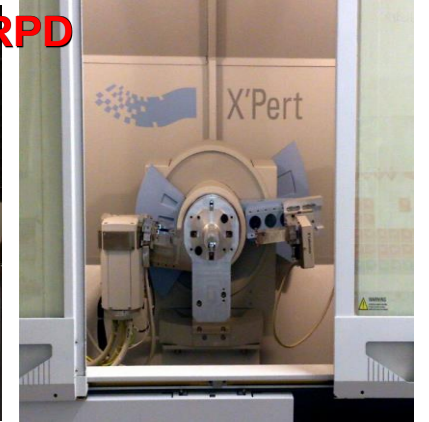
Fornace



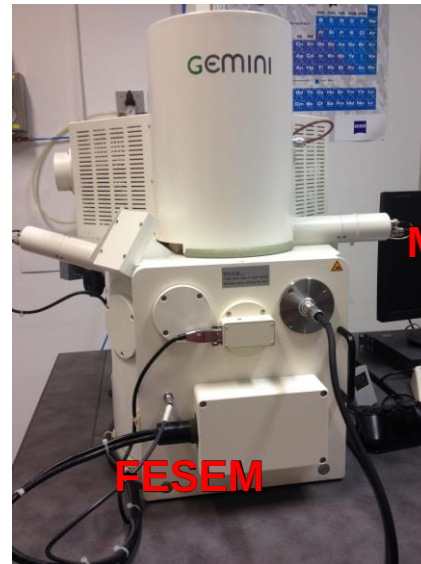
SEM



XRPD



X'Pert



FESEM

Micro-TC



Università del Triestino

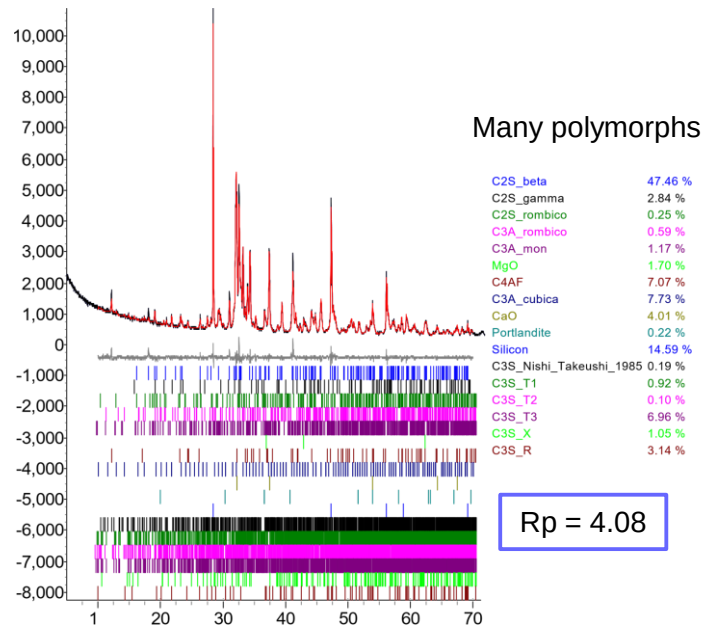
Il clincker ingegnerizzati con nCaO nei nostri labs



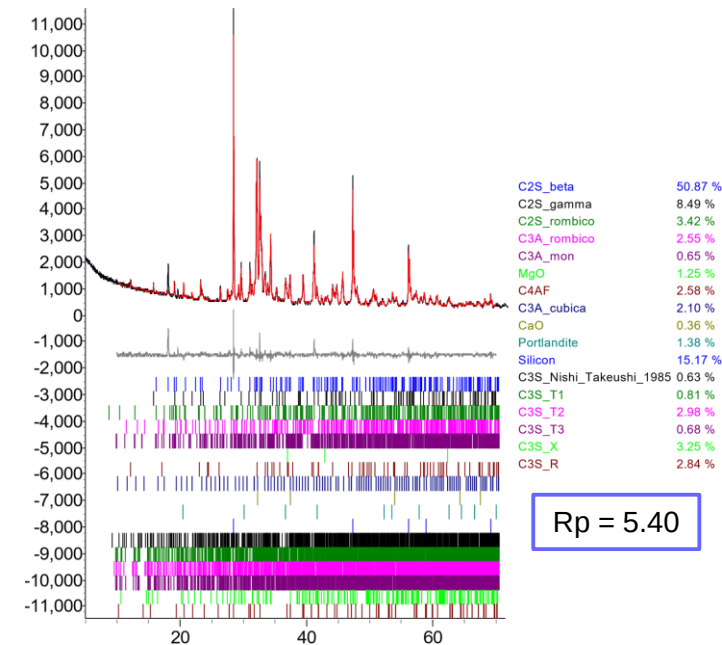
1250°C

Nano – Ca(OH)₂

Without nano – addition



With nano – addition



Il clincker ingegnerizzati con nCaO nei nostri labs

1250°C

Nano – Ca(OH)₂

Without nano – addition

Phase	wt%	Rp
C2S	42.0(2)	4.08
C3S	7.0(3)	

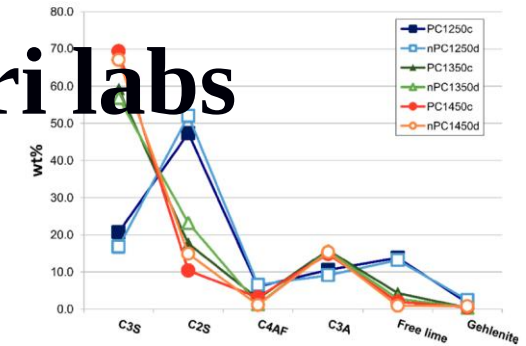
With nano – addition

Phase	wt%	Rp
C2S	46.2(2)	5.40
C3S	5.3(3)	

Aumento della cuocibilità del clinker

Miglioramento produttività impianto

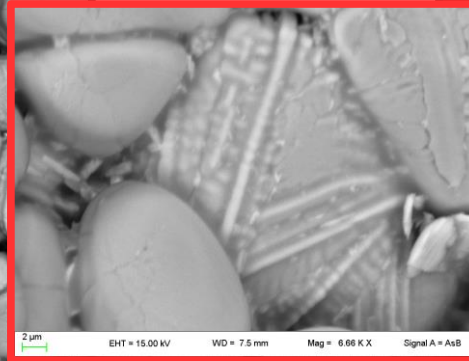
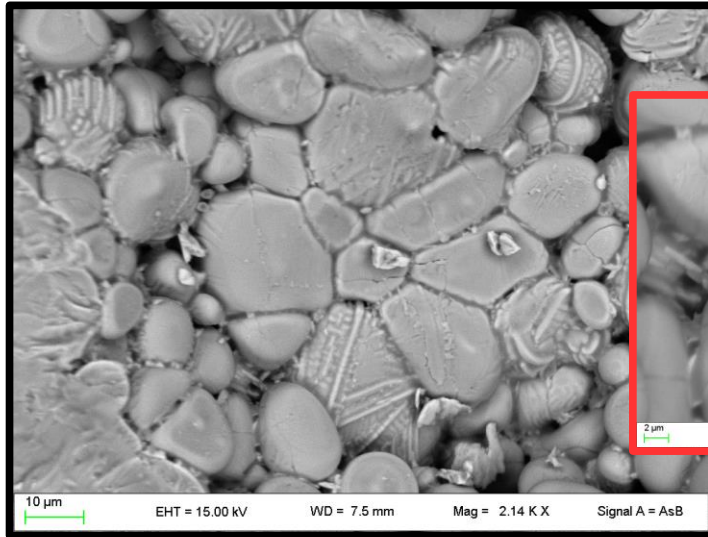
Minori emission di CO₂



Il clincker ingegnerizzati con nCaO nei nostri labs

1350

Nano – $\text{Ca}(\text{OH})_2$



CaO enrichment



EDX chemical map

nCaO aumenta le superfici reattive favorisce
 $\text{C}_2\text{S} + \text{CaO}$

I clincker ingegnerizzati con nCaO al sincrotrone

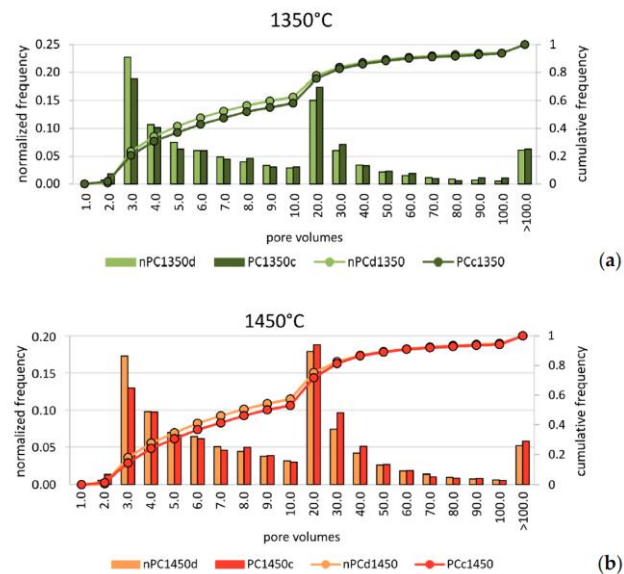
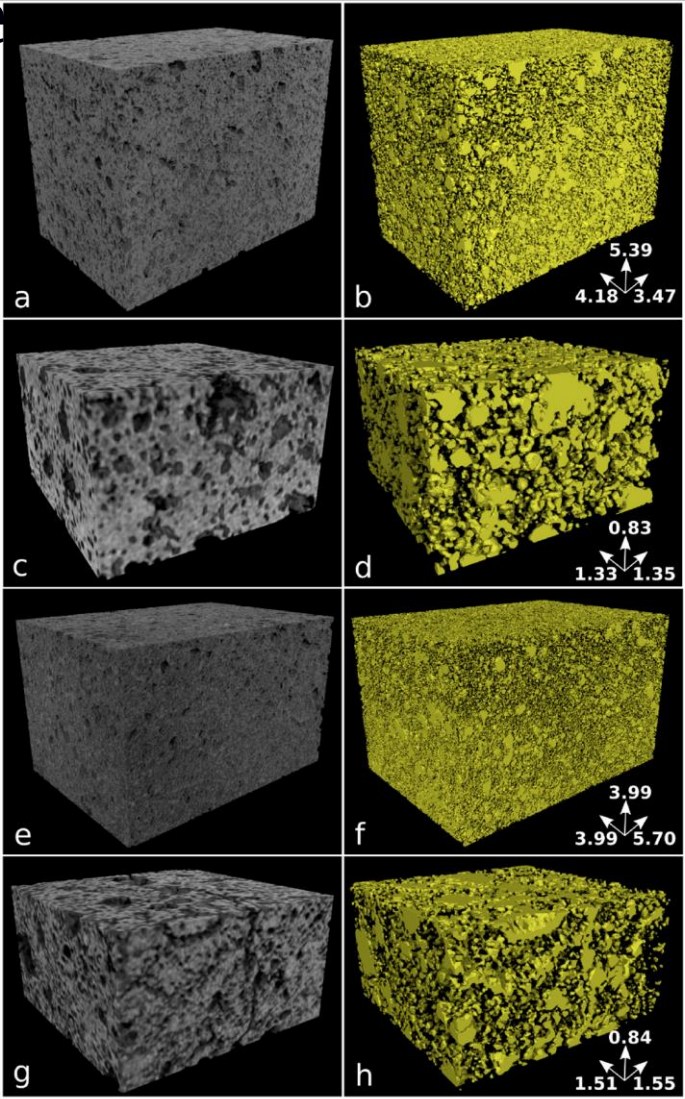


Table 2. Results obtained by applying the Basic Analysis module of the Pore3D software library extracted from the laboratory X-ray μ CT data.

Sample	Volume (voxels)	Voxel Size (μm)	Volume (mm^3)	Pores Thresholds	Eulerian Characteristics (mm^{-3})	Internal Mean Curvature (mm^{-2})	Specific Surface Area (mm^{-1})	Pore Fraction
nPC1450d	$946 \times 610 \times 734$	5.7	52.95	91	573	712	22.51	0.24
PC1450c	$271 \times 267 \times 167$	5.0	1.51	93	885	711	20.72	0.20
nPC1350d	$700 \times 700 \times 1000$	5.7	61.25	91	2048	711	16.77	0.16
PC1350c	$303 \times 168 \times 311$	5.0	1.98	76	1013	560	18.09	0.23



Diminuzione porosità con nCaO

Figure 7. Volume renderings of the extracted VOIs from reconstructed images (left column) and corresponding iso-surface renderings of the pore phase in yellow (right column) for samples nPC1450d (a,b), PC1450c (c,d), nPC1350d (e,f), and PC1350c (g,h). Samples nPC1450d and nPC1350d have been reconstructed with an isotropic voxel size of 5.7 μm , while an isotropic voxel size of 5.0 μm has been used for samples PC1450c and PC1350c. The size of analyzed VOIs is reported in Table 2. The dimension of the sample edges is reported in mm.

Conclusioni ... ad oggi

uso dei nano CaO nei cementi ... la nostra esperienza ci dice:

Vantaggi:

- 1) Uso di CaO invece di CaCO_3 riduce emissioni di CO_2 in partenza perché è già un materiale decarbonatato.. Non occorre la reazione $\text{CaCO}_3 = \text{CaO} + \text{CO}_2$
- 2) I nano-materiali costituiscono un ottimo catalizzatore di reazione nella produzione clinker favoriscono la formazione di belite a più basse temperature, riduzione di energia e produzione CO_2
- 3) Diminuzione della calce libera ...del 35% a 1350 °C e del 55% a 1450 e quindi un miglioramento della cuocibilità (CaO residua produce della porosità secondaria indesiderata)
- 4) Importante cambio nella porosità



Article

Effect of the Nano- $\text{Ca}(\text{OH})_2$ Addition on the Portland Clinker Cooking Efficiency

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Nel libro dei desideri....

Cementi nano-ingegnerizzati basati su materiali da
Costruction and Demolition Waste C&DW

**Stretta collaborazione
Con esperti di diverse
discipline e ...
con imprese del
settore**

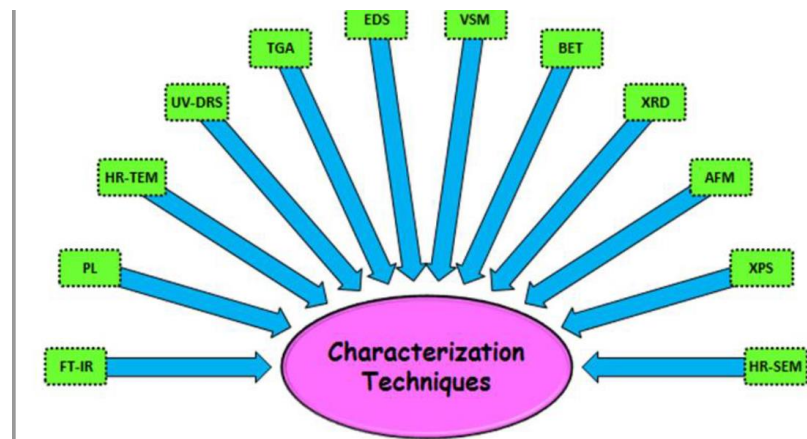


Figure 5. Characterization techniques for NPs.



Grazie per l'attenzione

